





# NSAT-2024

CLASS – X (Mental Ability, Biology, Physics & Chemistry)

(Class X Moving to XI-PCB)

## NARAYANA SCHOLASTIC APTITUDE TEST (NSAT)

## SAMPLE PAPER

Time: 1:00 Hr.

Maximum marks: 160

#### **IMPORTANT INSTRUCTIONS:**

- The test Booklet consists of 40 questions. The maximum marks are 160. 1.
- 2. There are five parts in the question paper of MAT (Q. No. 1 to 8) Biology (Q. No. 9 to 20), Physics (Q. No. 21 to 30) & Chemistry (Q. No. 31 to 40) having 40 questions. Each question is allotted +4 (four) marks for each correct response & -1 for each incorrect answer
- 3. Mark only one correct answer out of four alternatives.
- Use Blue/Black Ball Point Pen only for writing particulars/marking. 4.
- 5. Use of Calculator is not allowed.
- 6. Dark the circle in the space provided only.
- Use of white fluid or any other material which damage the answer sheet, is not permissible on the 7. Answer Sheet. ASPIRE • ASSESS • ACHIEVE

### **TO BE FILLED IN CAPITAL LETTERS**

NAME OF THE STUDENT : \_\_\_\_\_

FATHER'S NAME : \_\_\_\_

CONTACT NUMBERS: SCHOOL NAME :

ROLL NO. :\_\_\_\_\_\_\_\_TEST CENTRE : \_\_\_\_\_\_

I have read all the instructions and shall abide by them

I have verified all the information filled in by the Candidate

Signature of the Candidate

Signature of the Invigilator

#### **Class X TO XI-PCB**



EDUCATION IS INTEGRAL FOR GROWTH AND DEVELOPMENT

Education is integral for the growth and development of an individual. The expectation from an educational institute is always about making the society better for all and to bring out one's true Potential in the service of mankind.

At Narayana, we believe that a student's education is complete only when we are able to contribute towards his/her overall development besides imparting knowledge based and career oriented training.

With an aim to provide top of the league training to students to excel in every sphere of their lives, Narayana Group has been focusing on result oriented inputs.

Narayana's courses have been designed to cater to all the needs of the aspirants to help them excel in various competitive as well as Board examinations. Innovative strategies and techniques adopted in our centres keep our students abreast of the ever-changing pattern of top level Engineering/Medical Entrance Exams. As a result, Narayana's timetested learning formulae are percolating to far-flung corners of India to benefit students from all backgrounds.

"Footprints on the sands of time are not made by sitting down". Today we rededicate the last 4 decades of our success to your dreams. I wish all our students a very successful academic year ahead.

**Dr. P. NARAYANA** Founder, Narayana Group

NARAYANA IIT/NEET/FOUNDATIONS ACADEMY

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	<ul><li>(A) Stem, flowers and fruits</li><li>(C) Stem, roots and flowers</li></ul>	<ul><li>(B) Stem, leaves and flowers</li><li>(D) Stem, roots and leaves</li></ul>	
10.	Which among the following diseases is not sexually transmitted?		
	(A) Syphilis	(B) Hepatitis-A	
	(C) AIDS	(D) Gonorrhea	
11.	The part of seed that contains the futur	re plant is called the	
	(A) Cotyledons	(B) Seed coat	
	(C) Germ cells	(D) Embryo	
12.	Glycolysis occurs in which part of the	cell?	
	(A) Cytoplasm	(B) Nucleus	
	(C) Mitochondria	(D) Chloroplast	
	(B) They store carbohydrates in form	ofstarch	
	<ul><li>(C) They convert carbon dioxide and v</li><li>(D) They form the first trophic level in</li></ul>	water into carbohydrates in the absence of sunlight. In the food chain.	
14.	<ul><li>(C) They convert carbon dioxide and v</li><li>(D) They form the first trophic level ir</li><li>Along the path of the vas deferens the</li></ul>	water into carbohydrates in the absence of sunlight. In the food chain. secretions of which gland provide nutrition to the sperms	
4.	<ul><li>(C) They convert carbon dioxide and v</li><li>(D) They form the first trophic level ir</li><li>Along the path of the vas deferens the</li><li>(A) Bulbourethral gland</li></ul>	water into carbohydrates in the absence of sunlight. In the food chain. secretions of which gland provide nutrition to the sperms (B) Seminal vesicles	
4.	<ul> <li>(C) They convert carbon dioxide and v</li> <li>(D) They form the first trophic level in</li> <li>Along the path of the vas deferens the</li> <li>(A) Bulbourethral gland</li> <li>(C) Scrotum</li> </ul>	water into carbohydrates in the absence of sunlight. In the food chain. secretions of which gland provide nutrition to the sperms (B) Seminal vesicles (D) Urinary bladder	
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16.	Movement of sunflower in accordance with the path of the sun is due to		
	(A) Chemotropism	(B) Geotropism	
	(C) Phototropism	(D) Hydrotropism	
17.	Breathing is controlled by which part of the brain.		
	(A) Cerebrum	(B) Cerebellum	
	(C) Hypothalamus	(D) Medulla oblongata	
18.	Which of the following endocrin	ne glands does not exist in pairs?	
	(A) Testes	(B) Adrenal	
	(C) Pituitary	(D) Ovary	
19.	Length of pollen tube depends of	on the distance between	
	(A) Pollen grain and upper surfa	ace of stigma.	
	(B) Pollen grain on upper surfac	ce of stigma and ovule.	
	(C) Pollen grain in anther and u	pper surface of stigma.	
	(D) Upper surface of stigma and	l lower part of style.	
20.	Offspring formed as a result of s	sexual reproduction exhibit more variations because	
	(A) Sexual reproduction is a len	gthy process.	
	(B) Genetic material comes from	n two parents of the same species.	
	(C) Genetic material comes from	n two parents of different species.	
	(D) Genetic material comes from	m many parents.	
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## PHYSICS

- A person has near point 60 cm. What power should corrective less have to allow to see an object clearly at a distance of 20 cm.
  (A) +3.33D
  (B) -3.33D
  (C) +2.2D
  (D) -2.2.2D
- 22. A battery of EMF 2V and internal resistance  $0.5\Omega$  is connected across a resistance of 9.5Ω. How many electrons cross through cross-section of resistance per second? (A)  $2 \times 10^{-18}$  (B)  $3 \times 10^{-18}$  (C)  $1.25 \times 10^{18}$  (D)  $1.75 \times 10^{15}$
- 23. In the given figure, when galvanometer shows no deflection, the current (in ampere) flowing through 5 $\Omega$  resistance will be



24. In the arrangement of resistances shown below, the effective resistance between point A and B is

		$1 \Omega 1 \Omega 1 \Omega B$ $1 \Omega 1 \Omega 0$ $2\Omega 1 \Omega 0$ $3\Omega 0$	
(A) 20Ω	(B) 30 Ω	(C) 90Ω	(D) 110Ω

25. If f is the focal length for mirror, u is the distance of the object from mirror & v is the distance of the image from the mirror, then which of the following relation is true

(A) 
$$f = \frac{uv}{u+v}$$
 (B)  $u = \frac{f+v}{fv}$  (C)  $v = \frac{f+v}{fu}$  (D) None of these

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26.	6. A ray of light incident normally on a plane mirror. The angle of reflection will be		
	(A) $0^0$ (B) $90^0$		
	(C) Will not be reflected (D) None of these		
27.	The position and nature of image formed when an object of size 1 cm is placed at a distance of	f	
	15 cm from a concave mirror of focal length 10 cm is		
	(A) $-30$ cm, real and inverted (B) $-20$ cm virtual and inverted		
	(C) $-40$ cm, real and inverted (D) $-25$ cm virtual and enlarged		
28.	An object placed 10 cm in front of a lens has an image 20 cm behind the lens. What is the pow	ver	
	of the lens (in diopters)	•	
	(A) 1.5 (B) 3.0 (C) $-15.0$ (D) $+15.0$		
20			
29.	A bird is flying at a height of 3.6 m above the surface of water and a fish is in the water at a		
	depth of 1.2 m. The apparent height of bird to the fish is $\left( \mu_{water} = \frac{4}{3} \right)$		
	(A) 3.9 m (B) 4.8 m (C) 5.2 m (D) 6.0 m		
30.	A concave lens of suitable focal length is used for correcting a –		
	(A) Myopic eye		
	(C) Both (A) & (B) (D) Neither (A) nor (B)		
	ASPIRE · ASSESS · ACHIEVE		
	Space for rough work		
NAD	Α.Υ.Α.Ν.Α. Π.Υ./ΝΕΕΥ/ΕΩΙΙΝΙΝΑΤΙΩΝΙς Α.Ο.Α.ΝΕΜΙΥ	_	

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31.	If H <sup>+</sup> ion concentration of a solution is (A) increases by 1 (C) decreases by 1	increased 10 times its pH will (B) remains unchanged (D) increases by 10	
32.	Identify true and false statement ? (1) The relationship between pH and hy (2) By the dilution of an acidic solution (A) TF (B) FT	ydrogen ion concentration is inverse one. n concentration of H <sup>+</sup> ion increases continuously (C) FF (D) TT	
33	A metal compound 'A' react with dilute hydrochloric acid to produce effervescence. The gas evolved extinguishes a burning candle. The balanced chemical equation for the above reaction is (A) NaHCO <sub>3</sub> + HCl (dilute) $\rightarrow$ NaCl + H <sub>2</sub> O + CO <sub>2</sub> (B) NaOH + HCl (dilute) $\rightarrow$ NaCl + H <sub>2</sub> O (C) CuSO <sub>4</sub> + 2HCl $\rightarrow$ CuCl <sub>2</sub> + H <sub>2</sub> SO <sub>4</sub> (D) None of these		
34.	An oxide of a non-metal has the follow (i) It acts both as proton donor or accept (ii) It readily reacts with basic and acid (iii) It oxidises iron at its boiling point The oxide is (A) H <sub>2</sub> O (B) SO <sub>2</sub>	ing properties: btor lic oxides OLASTIC APTITUDE TEST (C) NO <sub>2</sub> (D) CO <sub>2</sub>	
35.	When sodium bisulphite reacts with hy $(A)$ NaCl HaO and SO:	drochloric acid, the products formed are : (B) Na <sub>2</sub> SO <sub>4</sub> and NaCl	
	(C) NaCl and $H_2S$	(D) No reaction takes place	
36.	Which of the following statements is $\Pi$ (A) The conjugate base of $H_2PO_4^-$ is $H$ (B) The pH of 1 M HCl is 0. (C) $H_3PO_3$ is a tribasic acid. (D) The concentration of $H^+$ ions in put	NCORRECT ? PO4 <sup>2-</sup> . re water is $10^{-7}$ mol L <sup>-1</sup> at 298 K.	
37.	$Pb(NO_3)_2 + 2KI \rightarrow PbI_2 + 2KNO_3.$		
	(A) displacement reaction (C) combination	<ul><li>(B) double displacement reaction</li><li>(D) decomposition</li></ul>	

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- 38. Which one of the following acids has two replaceable hydrogen atoms?(A) Formic acid (B) Acetic acid (C) Sulphuric acid (D) Phosphoric acid
- 39. Which of the following metals is present in the anode mud during the electrolytic refining of copper?(A) Sodium (B) Aluminium (C) Gold (D) Iron
- 40. An aluminium strip is kept immersed in freshly prepared ferrous sulphate solution taken in a test tube, the change observed is that
  - (A) Green solution slowly turns brown
  - (B) Lower end of test tube become slightly warm
  - (C) A colourless gas with the smell of burning sulphur is observed
  - (D) Light green solution changes to blue.



Space for rough work

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